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#### **Key indicators**

Single-crystal X-ray study T = 93 K Mean  $\sigma$ (C–C) = 0.004 Å R factor = 0.054 wR factor = 0.157 Data-to-parameter ratio = 16.3

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. Received 25 February 2007 Accepted 3 April 2007

# Ammonium bis{3-anilinocarbonyl-1-[(5-chloro-2-oxidophenyl)diazenyl]-2-naphtholato}ferrate(III) methanol solvate

The title compound,  $(NH_4)$ [Fe(C<sub>23</sub>H<sub>24</sub>ClN<sub>3</sub>O<sub>3</sub>)<sub>2</sub>]·CH<sub>4</sub>O, is a charge-control agent used in electrophotography. The central Fe<sup>III</sup> atom is coordinated by four O atoms and two N atoms of two symmetry-independent ligands. The metallic complex has no crystallographic symmetry. The structure features hydrogen-bonded double chains, which run along the *a* axis.

## Comment

The title compound, (I), is a methanol-solvated, azo-iron complex used widely as a charge-control agent (CCA) of the negative type in electrophotography (Tanaka, 1995). CCAs are usually added to toners to create a desired charge level and polarity (Nash *et al.*, 2001). However, the charge-control mechanism of CCA is not yet fully understood. An attempt was therefore made to determine the crystal structure as a step towards elucidation of the mechanism. This paper deals with the structure of the methanol-solvated complex; the acetone-solvated complex (Mizuguchi *et al.*, 2007) follows this publication.

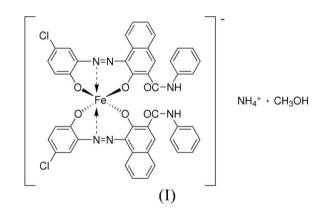


Fig. 1 shows the molecular structure of (I). The anion complex has no crystallographic symmetry, though inversion symmetry (which is not possible) has been implied in literature (for example, Tanaka, 1995). Fig. 2 illustrates the packing arrangement in (I), showing ions related by inversion centers and the hydrogen-bond network (Table 1) linking the metal complexes in double chains running along the a axis.

# **Experimental**

Compound (I) was prepared according to the methods previously reported (Yasumatsu *et al.*, 2006). Single crystals of (I) were recrystallized from a methanol solution. After 48 h, a number of black crystals were obtained in the form of blocks.

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# metal-organic papers

#### Crystal data

 $\begin{array}{l} (\mathrm{NH}_4)[\mathrm{Fe}(\mathrm{C}_{23}\mathrm{H}_{24}\mathrm{ClN}_3\mathrm{O}_3)_2]\cdot\mathrm{CH}_4\mathrm{O}\\ M_r = 937.58\\ \mathrm{Triclinic}, \ P\overline{1}\\ a = 10.1107 \ (\mathrm{4}) \ \mathrm{\mathring{A}}\\ b = 14.1645 \ (\mathrm{6}) \ \mathrm{\mathring{A}}\\ c = 15.3877 \ (7) \ \mathrm{\mathring{A}}\\ \alpha = 103.2240 \ (11)^\circ\\ \beta = 102.8270 \ (12)^\circ \end{array}$ 

### Data collection

Rigaku R-AXIS RAPID diffractometer Absorption correction: multi-scan (ABSCOR; Higashi, 1995)  $T_{\rm min} = 0.896, T_{\rm max} = 0.940$ 

## Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.054$  $wR(F^2) = 0.157$ S = 1.009404 reflections

Table 1	_	
Hydrogen-bond geometry	(Å,	°).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$N7-H1A\cdots O1$	1.06	1.69	2.747 (3)	174
$N7-H1B\cdots O6^{i}$	0.95	1.96	2.910 (3)	178
$N7 - H1C \cdot \cdot \cdot O7$	0.92	1.84	2.748 (4)	173
$N7 - H1D \cdots O3^{i}$	0.84	1.92	2.753 (3)	171
O7−H7O···O6 <sup>ii</sup>	0.84	1.89	2.709 (3)	166
N3−H3 <i>N</i> ···O2	0.88	2.09	2.675 (3)	124
N6−H6 <i>N</i> ···O5	0.88	1.95	2.653 (2)	136

 $\nu = 94.0860 \ (12)^{\circ}$ 

Mo  $K\alpha$  radiation  $\mu = 0.56 \text{ mm}^{-1}$ 

T = 93.1 K

 $R_{\rm int} = 0.054$ 

578 parameters

 $\Delta \rho_{\rm max} = 0.77$  e Å<sup>-</sup>

 $\Delta \rho_{\rm min} = -0.92 \text{ e} \text{ Å}^{-3}$ 

Z = 2

V = 2074.39 (15) Å<sup>3</sup>

 $0.20 \times 0.20 \times 0.10 \; \mathrm{mm}$ 

32362 measured reflections

9404 independent reflections 6242 reflections with  $F^2 > 2\sigma(F^2)$ 

H-atom parameters constrained

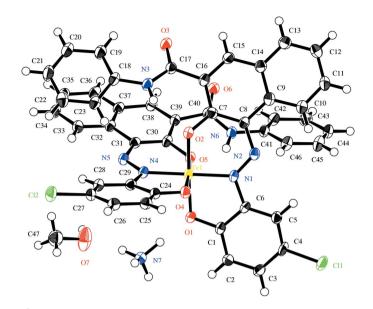
Symmetry codes: (i) x - 1, y, z; (ii) -x + 1, -y + 1, -z + 1.

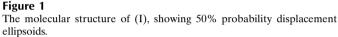
The four H atoms of the ammonium cation were found in difference maps and fixed in position during the least-squares refinement with  $U_{iso}(H) = 1.2U_{eq}(N)$ . All remaining H atoms were placed in geometrically idealized positions and constrained to ride on their parent atoms, with C-H = 0.95 and 0.98 Å, N-H = 0.88 Å, O-H = 0.84 Å, and  $U_{iso}(H) = 1.2U_{eq}(\text{parent atom})$ .

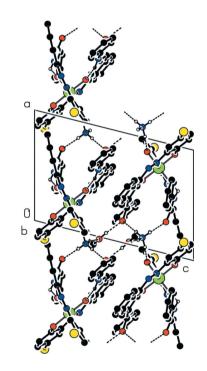
Data collection: *PROCESS-AUTO* (Rigaku, 1998); cell refinement: *PROCESS-AUTO*; data reduction: *CrystalStructure* (Rigaku/ MSC, 2006); program(s) used to solve structure: *SIR2004* (Burla *et al.*, 2003); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPIII* (Burnett & Johnson, 1996); software used to prepare material for publication: *CrystalStructure*.

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# Figure 2

The packing arrangement in (I) showing a double chain. The H atoms, except for the ammonium cation and methanol solvent molecule, have been omitted for clarity. Dashed lines indicate hydrogen bonds.

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